Bioinformation 19(4): 403-406 (2023)

©Biomedical Informatics (2023)

OPEN ACCESS GOLD

DOI: 10.6026/97320630019403



Received April 1, 2023; Revised April 30, 2023; Accepted April 30, 2023, Published April 30, 2023

Declaration on Publication Ethics:

The author's state that they adhere with COPE guidelines on publishing ethics as described elsewhere at https://publicationethics.org/. The authors also undertake that they are not associated with any other third party (governmental or non-governmental agencies) linking with any form of unethical issues connecting to this publication. The authors also declare that they are not withholding any information that is misleading to the publisher in regard to this article.

Declaration on official E-mail:

The corresponding author declares that lifetime official e-mail from their institution is not available for all authors

License statement:

This is an Open Access article which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly credited. This is distributed under the terms of the Creative Commons Attribution License

Comments from readers:

Articles published in BIOINFORMATION are open for relevant post publication comments and criticisms, which will be published immediately linking to the original article without open access charges. Comments should be concise, coherent and critical in less than 1000 words.

> **Edited by P Kangueane** Citation: Govindaraj et al. Bioinformation 19(4): 403-406 (2023)

Green synthesis, characterization, and antibacterial activity of Citrus lanatus based silver nanoparticles

Aravinthrajkumar Govindaraj, Mahesh Ramakrishnan, Rajeshkumar Shanmugam & SP Saravana Dinesh

Department of Orthodontics, Saveetha Dental College and Hospitals, Chennai-600077; *Corresponding author

Affiliation URL:

https://saveethadental.com/

Author contacts:

Aravinthrajkumar Govindaraj - E-mail: draravinthrajkumar.sav@gmail.com; Phone: +91 8072332799; ORCID: 0000-0001-9485-0302 Mahesh Ramakrishnan - E-mail: mahesh@saveetha.com; ORCID: 0000-0002-6367-6855 Rajeshkumar Shanmugam - E-mail: rajeshkumars.sdc@saveetha.com; ORCID: 0000-0001-7059-8894 SP Saravana Dinesh - E-mail: saravanadinesh@saveetha.com; Phone: +91 9884115197; ORCID: 0000-0001-5302-7483

Abstract:

Due to their enhanced and unique physicochemical characteristics, such as their minimal dimensions, large surface area compared to their mass and increased reactivity, nanomaterials hold promise in the field of antibacterial therapy. The aim of this study is to evaluate the Bioinformation 19(4): 403-406 (2023)

method for Green Synthesis, Characterization of *Citrus lanatus* pulp, rind, and seed-based Silver Nanoparticles, and study its antibacterial activity against common oral microorganisms. Green synthesis of nanoparticles was formulated using the extract from the rind, seed, and pulp of *Citrullus lanatus* and silver nitrate extract. The extract was stirred in a magnetic stirrer for 24 hours, and centrifugation was done at 8000 rpm for 10 minutes. The synthesized nanoparticles were characterized and utilized for the antibacterial study. The characterization was done using UV-visible light spectroscopy, and antibacterial activity was assessed using the well diffusion method and measuring the zone of inhibition. The nanoparticles synthesized were characterized using UV-visible spectroscopy, and the spectroscopic analysis confirmed the formation of silver particles in this study. At 450 nm, a sharp peak was observed, which correlated to the SPR band of the particles. The results showed that the zone of inhibition assessed was comparable to the standard. From this study it can be concluded that silver nanoparticles (AgNPs) produced by green synthesis from *Citrullus lanatus* showed significant antibacterial against common oral microflora.

Background

Nanotechnology is a branch of science that deals with nanoparticles, or things that are smaller than a nanometer (NPs). Nanomaterials are tiny solid particles with a size between one and one hundred nanometers. Due to their enhanced and unique physicochemical characteristics, such as their minimal dimensions, large surface area compared to their mass and increased reactivity, nanomaterials hold promise in the field of antibacterial therapy. [1] The availability of a wide variety of metabolites with strong reduction potentials, global distribution, safe handling, minimal waste and energy costs, large and accessible reserves, and widespread use of plant extracts are only a few of the reasons why they are so popular. [2] Due to several features, including a variable surface area-to-volume ratio that is useful in a variety of biological and technological applications, silver nanoparticles (AgNPs) have received commercial attention. Their biological activity in relation to antibacterial, anticancer, and antioxidant effects in medical settings have been assessed in various studies.[3-5] In comparison to numerous antibacterial compounds, Silver(Ag) NPs have demonstrated a very high antimicrobial action with good biocompatibility. NPs have also been employed as the primary component of antibacterial nan ocomposites based on inorganic and polymeric materials.[6] There are physical, chemical, and biological ways to make silver nanoparticles (NPs). In the top-down approach, bulk metals are mechanically ground, and the resulting nanosized metal particles are then stabilized by the addition of colloidal stabilizers. Contrarily, the bottom-up approaches use sono decomposition, electrochemical techniques, and metal reduction. The simplest approach involves reducing the metal salts AgBF₄ chemically with NaBH4 in water. The NPs that are produced in this manner range in size from 3 to 40 nm.[7-9] Utilizing microorganisms like fungi, yeasts (eukaryotes), bacteria, and actinomycetes (prokaryotes), using plants and plant extracts, or using templates like membranes, viruses' DNA, and diatoms are all examples of green synthesis.[10] In this original article, we are going to see the method for green synthesis, the characterization of Citrus lanatus pulp, rind, and seed-based silver nanoparticles, and study their antibacterial activity against common oral microorganisms.

Materials and Methodology:

The study was conducted in the department of pharmacology and nanotechnology at the university after getting approval from the institutional scientific committee and the scientific review board. This is an in vitro study that was conducted for the green synthesis of silver nanoparticles from the rind, pulp, and seed extract of *Citrullus lanatus.*

Preparation of Citrullus lanatus extracts:

A fresh-cut pulp of *Citrullus lanatus* was cut using a sterile knife. The seeds were removed from the cut pieces, and the pulp was crushed using a mortar and pestle to obtain 100 ml of pure extract. The obtained watermelon juice was filtered using sieves, and secondary filtration was done using Whatman grade 1 filter paper. To obtain the extract of the seeds and rind of *Citrullus lanatus*, the seed and rind were cut and dried in a hot air oven. The dried seed and rind of *Citrullus lanatus* were powdered, and the extract was prepared using distilled water. The amount of material used in the extract's synthesis can be changed depending on the amount of extracted particles required.



Figure 1: Silver nitrate (AgNO₃) mixed with *Citrullus lanatus* extract.

Synthesis of Nanoparticles:

Green synthesis of silver nitrate nanoparticles weighing 0.0169 g dissolved in 90 mL distilled water to produce a pure extract of silver nitrate. Fresh *Citrullus lanatus* extract was added to that 10 mL and stirred in a magnetic stirrer for 24 hours **[Figure 1]**. After which, centrifugation was done at 8000 rpm for 10 minutes. The supernatant was discarded, and the pellet was stored in an airtight Eppendorf tube at 4 degrees Celsius. The synthesized nanoparticles were characterized and utilized for the antibacterial study. The quantity of the material used in the synthesis of nanoparticles and the duration taken to prepare them can be altered and will change,

Bioinformation 19(4): 403-406 (2023)

respectively, depending on the quantity of the nanoparticle required.

Characterization of nanoparticles:

UV-vis spectroscopy was used to assess the produced nanoparticles' UV-vis absorption peak. The chosen scanning range is 330-660 nm. The transformation of Citrullus lanatus with silver nitrate solution into AgNPs is necessary for UV-vis spectroscopic investigation. The appearance of a shift in hue will signal the creation of NPs.

Antibacterial activity:

The antibacterial activity of the extracted *Citrullus lanatus* based silver nanoparticles was assessed against common oral bacteria such as *S. aureus, S. mutans, C. albicans, E. faecalis, and Lactobacillus sp.* For this experiment, MHA agar was used to identify the zone of inhibition. Prepared Muller-Hinton agar was sterilized for 45 minutes at 120 lbs. Sterilized plates were filled with the medium, which was then left to solidify. The test organisms were swabbed after the wells were cut with the well cutter. Three different sources of nanoparticles were added, keeping chlorhexidine as the standard, and the plates were incubated for 24 hours at 37 °C. The zone of inhibition was assessed following the incubation period.

Results:

Ultraviolet-visible spectroscopy:

The surface plasmon resonance (SPR) shows an unusual occurrence of noble metal nanoparticles that generate intense electromagnetic fields on the surface of the particles, and thus the radioactive properties such as absorption and scattering are increased. Hence, UV spectroscopic analysis confirmed the formation of silver particles in this study. At 450 nm, a sharp peak was observed, which correlated to the SPR band of the particles.

Antibacterial activity:

The antibacterial activity of the synthesized nanoparticles was studied by assessing the zone of inhibition (Figure 2).



Figure 2: Zone of Inhibition; A - Antimicrobial effect of *Citrullus lanatus* based silver nanoparticles on *Lactobacillus sp.;* **B -** Antimicrobial effect of *Citrullus lanatus* based silver nanoparticles

on *S. mutans.;* **C** - Antimicrobial effect of *Citrullus lanatus* based silver nanoparticles on *S. aureus.;* **D** - Antimicrobial effect of *Citrullus lanatus* based silver nanoparticles on *E. faecalis.;* **E** - Antimicrobial effect of *Citrullus lanatus* based silver nanoparticles on *C. albicans.*

The zone of inhibition measured using the vernier caliper is given in the table below **[Table 1]**. The zone of inhibition showed that there was significant antibacterial activity against all the bacteria. The antibacterial activity was increased in seed NPs when compared to that of pulp and rind NPs, and the antibacterial activity was significantly higher against *Staphylococcus aureus* than the other bacteria. The antibacterial activity of all the nanoparticles was comparable to that of the standard, chlorhexidine.

Table 1: Zone of Inhibition

| Bacteria sp. | Pulp NPs | Rind NPs | Seed NPs | Standard |
|-------------------|----------|----------|----------|----------|
| Lactobacillus sp. | 9 | 10 | 14 | 18 |
| S.mutans | 14 | 16 | 18 | 22 |
| S.aureus | 22 | 26 | 29 | 32 |
| E.fecalis | 14 | 16 | 18 | 22 |
| C.albicans | 12 | 10 | 11 | 16 |

Discussion:

The findings of this study demonstrated that Ag nanoparticles, green synthesized from *Citrullus lanatus*, have a significant antibacterial effect against common oral bacteria like S. mutans, S. aureus, C. albicans, Lactobacillus sp., and E. fecalis. AgNPs are a promising system with key characteristics including antibacterial, anti-inflammatory, and anticancer action, according to research. They may also be used as carriers in sustained drug delivery. They said that more research was needed to fully understand some elements of the processes by which AgNPs work, as well as some significant toxicological issues brought on by the usage of this technology. [11,12] Researchers looked at the antibacterial effects of AgNPs added to dental materials such as composite resin, endodontic materials, acrylic resin, and implants. Additionally, numerous studies have demonstrated that silver, in its nanoparticulated form, has an inhibitory effect against a variety of bacteria and fungi, including S. mutans, C. albicans, P. aeruginosa, E. faecalis, and S. aureus, among others. This could reduce the risk of secondary caries, fungal infections, failed endodontic treatments, and dental implants.[13,14] The antibacterial activity of Citrullus lanatus has been studied earlier in various studies earlier. The study on the antibacterial activity of watermelon extract against oral microflora showed that there was a significant effect against Lactobacillus [15]. The activity of various components of Citrullus lanatus was studied in various articles and showed that all the components of it, including the rind, pulp, and seed, showed significant antibacterial activity against oral microflora like Streptococcus mutans, Staphylococcus aureus, Lactobacillus sp., Candida albicans, and E. fecalis.[16-18] The UV spectroscopy method was used to characterize the nanoparticles synthesized. Numerous studies have shown the use of a UV-Vis spectrophotometer to track the bio-reduction of silver ions in aqueous solutions. In these investigations, pure Ag+ ion reduction was often observed 3-5 hours after a tiny aliquot of the sample was diluted in distilled water. [19-22] Another study on the green synthesis of silver Ag

ISSN 0973-2063 (online) 0973-8894 (print)

Bioinformation 19(4): 403-406 (2023)

nanoparticles came to the conclusion that these particles might be used as antibacterial agents. The characterization research demonstrated that the particles created in nano dimensions would be just as useful in therapeutic formulations as antibiotics and other medications. **[23–25]**

Conclusion:

Thus, silver nanoparticles (AgNPs) produced by green synthesis from *Citrullus lanatus* rind, fruit, and seeds, showed significant antibacterial potential against common oral microflora. The antibacterial activity assessed by using the zone of inhibition was increased with seed extract AgNPs when compared with fruit and rind extract silver nanoparticles.

Clinical Significance:

By lowering the quantity of common oral bacteria that cause carious lesions, these newly created silver nanoparticles based on *Citrullus lanatus* can be employed in the indirect prevention and creation of new dental caries.

Limitations:

This in-vitro study compares the antibacterial effectiveness of silver nanoparticles derived from *Citrullus lanatus* to the industry standard of chlorhexidine. To confirm the study's findings, more research utilizing human clinical trials is essential.

Funding: There is no source of funding.

Acknowledgement:

The authors would like to thank Ms. Tharani Munuswamy for her contribution during the study.

Reference:

- [1] Yazdanian M *et al. Bioinorg Chem Appl* 2022 2022:2311910. [PMID: 35281331]
- [2] Souza JA *et al. Future Microbiol* 2018 13:345 [PMID: 29441824]

©Biomedical Informatics (2023)

- [3] Kumar A et al. J Sci Ind Res Vol.77(04) April 2018
- [4] Thomas R et al. 3 Biotech 2018 8. [PMID: 30221117]
- [5] Teimuri-Mofrad R *et al.* Nanochemistry Research 2017 2:8.
- [6] Noronha VT *et al. Dent Mater* 2017 33:1110. [PMID: 28779891]
- [7] Kaur P & Luthra R. SRM Journal of Research in Dental Sciences 2016 7:162.
- [8] Gaffet E *et al. Mater Charact* 1996 36:185.
- [9] Thirumalai Arasu V et al. J Bio Sci Res 2010 1:259.
- [10] Rafique M *et al. Artif Cells Nanomed Biotechnol* 2017 45:1272. [PMID: 27825269]
- [11] Rai M et al. Appl Microbiol Biotechnol 2014 98:1951. [PMID: 24407450]
- [12] Wei L *et al. Drug Discov Today* 2015 20:595. [PMID: 25543008]
- [13] Corrêa JM *et al. Int J Biomater* 2015 2015:485275. [PMID: 25667594]
- [14] Sarwar A et al. J Nanopart Res 2014 16:2517.
- [15] Govindaraj A & Dinesh SP. *Drug Invention* 2019.
- [16] Govindaraj A et al. Ann Dent Spec 2022 10:34.
- [17] Govindaraj A & Dinesh S. P. S. *Journal of Evolution of Medical and Dental Sciences.* 2020 Sep 14 9(37):2674.
- **[18]** Govindaraj A *et al. Journal of Complementary Medicine Research* 2022 11:230.
- [19] Chauhan RPS *et al. International Journal of Bioassays* (IJB) 2012.
- [20] Linga et al. Journal of Pharmaceutical Sciences and Research. 2011 3:1117.
- [21] Zargar M et al. Molecules 2011 16:6667. [PMID: 25134770]
- [22] Devi NN et al. International Journal of Pharmaceutical Sciences Review and Research. 2012 12:164.
- [23] Karpiński TM & Szkaradkiewicz AK. J Biol Earth Sci 2013 3:M21.
- [24] Al-Shahrani MA. Ann Med Health Sci Res 2019.
- [25] Berkowitz RJ. Pediatr Dent 2006 28:106 [PMID: 16708784]